

Preparation of activated carbon from olive mill solid residue

N Mameri,^{1*} F Aiouèche,¹ D Belhocine,¹ H Grib,¹ H Lounici,¹
DL Piron² and Y Yahiat³

¹Laboratory of Biotechnology, Ecole Nationale Polytechnique, 10 Avenue Pasteur, El Harrach, Algiers, Algeria

²Ecole Polytechnique de Montréal, Département Génie Physique et des Matériaux, PO Box 6079, Station Centre-ville, Montréal (Québec), Canada H3C 3A7

³Université Scientifique de Technologie Houari Boumediène (USTHB), BP 162, Bab ezzouar, Alger, Algeria

Abstract: A process was developed for producing high quality activated carbon from Algerian mill waste. The solid olive mill residue was carbonized at 800°C and physically activated with CO₂, air or steam. An optimum activation temperature of about 850°C was determined for all the activation agents used. Steam appeared to be the most efficient activator as compared with air and CO₂. An optimal activation time of about 5 h was then determined with steam as the optimum activation agent. The porous structure of the activated carbon was characterized by nitrogen adsorption at -196°C, and in all cases the surface areas, calculated by DR and BET methods, confirmed the production of a material with good microstructural characteristics and specific surfaces exceeding 1500 m²g⁻¹ for the carbon prepared by steam activation. Phenol adsorption isotherms gave the adsorption properties and the adsorption capacity of about 11.54 mg of phenol per gram of the activated carbon produced. The kinetics of the phenol adsorption onto the porous material was evaluated by means of two models: the external resistance model and the linear model. The second model appeared to constitute a more appropriate fit for the experimental data.

© 2000 Society of Chemical Industry

Keywords: activated carbon; olive mill waste; adsorption kinetics

1 INTRODUCTION

The Kabylia region in the north of Algeria is an important producer of olive oil, but this production is accompanied by the rejection of more than 10 000 tons of recalcitrant solid wastes, which causes a problem for agriculture. The inhibition properties of these wastes limits the efficiency of biological treatments,^{1–10} but this natural product has been utilized for agricultural purposes^{5,11} and to remove selected heavy metals from aqueous solutions.¹¹ The residual fly ash produced by coal combustion in thermoelectric power plants has also been mixed with vegetation water obtained from olive mills to obtain a material with high porosity and high specific surface area.¹⁵ The carbonized solid residue of olive mill products, called J carbon (0.6–0.8 mm), has been compared with commercial activated carbon in the treatment of industrial wastewater to remove NH₃ and total Organic Carbon as non-specific organics as well as with six specific leading organic pollutants.¹³ The J carbon appeared to be as efficient for treating industrial wastewater as other commercial activated carbons.

In the present study, Algerian activated carbons prepared from solid residue of olive mill products were

characterized and used to adsorb phenol in aqueous solution. Carbon preparation consisted of carbonization of the natural solid matter followed by physical activation (CO₂, steam or air).

2 EXPERIMENTAL

2.1 Solid residue olive mill characteristics

The olive mill solid residue used in this study was collected over 3 years (1994–1998) from the Tadmaït (Tizi Ouzou) olive oil plant and transported to the laboratory at 4°C. Before carbonization, the natural matter was washed in acidic solution (HCl, 1 M) for 1 h at 60°C. It was then rinsed with distilled water until a neutral pH was reached. Finally, it was dried at 60°C for 5 h. The average values of the main characteristics of the olive mill solid residue are given in Table I. All experiments were duplicated, and they showed differences of less than 10%.

2.2 Activated carbon preparation

The dried olive mill solid residue was first submitted to the carbonization operation, which consisted of introducing a mass *w* in a muffle furnace (Lyradia,

* Correspondence to: N Mameri, Ecole Polytechnique of Montreal, Department of Mechanical TTRL Laboratory, PO Box 6079, Station Centre-ville, Montreal (Quebec), Canada H3C 3A7

E-mail: nabil.mameri@courriel.polymtl.ca

(Received 4 January 2000; revised version received 21 February 2000; accepted 2 March 2000)

